

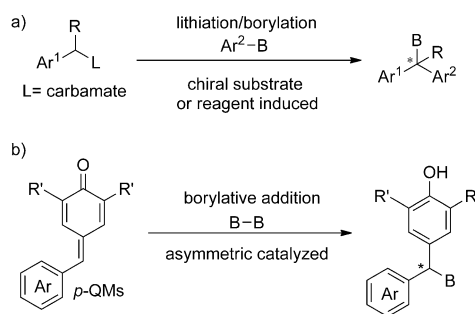
Copper-Catalyzed Enantioselective 1,6-Boration of *para*-Quinone Methides and Efficient Transformation of *gem*-Diarylmethine Boronates to Triarylmethanes

Yazhou Lou, Peng Cao, Tao Jia, Yongling Zhang, Min Wang, and Jian Liao*

Abstract: Presented is the first enantioselective copper-catalyzed 1,6-conjugate addition of bis(pinacolato)diboron to *para*-quinone methides. The reaction proceeds with excellent yields and good to excellent enantioselectivities, and provides an attractive approach to the construction of optically active *gem*-diarylmethine boronic esters. Additionally, the subsequent conversion of the derived potassium trifluoroborates into triarylmethanes with highly enantiospecificity was realized.

Enantiomerically pure organoboranes are interesting pharmaceutical molecules^[1] in medical chemistry and powerful building blocks^[2] in asymmetric synthesis. Chiral dibenzylic (*gem*-diarylmethine) boronates, as a class of potential precursors for important triarylmethanes^[3] and *gem*-diaryl hydrocarbons,^[4] has been employed by Aggarwal and co-workers^[5] in the synthesis of *gem*-diaryl-containing pharmaceuticals.^[5d–e] However, despite the importance of these compounds, the approaches to access nonracemic *gem*-diarylmethine boronates are still very scarce. Lithiation/borylation of benzylic carbamates^[6] is an exclusive route which relies on either a substrate- or reagent-controlled strategy. (Scheme 1a). For instance, Crudden and co-workers recently reported an elegant chiral bis(*i*Pr-oxazoline)-controlled protocol to obtain chiral nonracemic *gem*-diarylmethine boronic esters, albeit with substantial substrate restrictions.^[7] To our knowledge, to date, catalytic asymmetric approaches to access *gem*-diarylmethine boronic esters, bearing a chiral benzylic C–B bond, remain unexplored.

Transition-metal catalyzed^[8] or organocatalyzed^[9] asymmetric conjugate borations of electron-deficient olefins with diboron reagents, such as bis(pinacolato)diboron [B₂(pin)₂], represent a powerful and site-selective route to access chiral alkylboranes.^[10] In particular, the use of chiral copper catalysts has been highly successful in transforming a variety of β -aryl-substituted α,β -unsaturated compounds into benzylic boronic esters with excellent levels of enantiopurity.^[11]



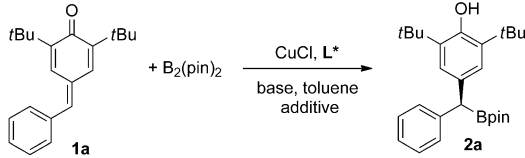
Scheme 1. Approaches to chiral *gem*-diarylmethine boronates.

para-Quinone methides (*p*-QMs),^[12] which are classified as electron-deficient alkenes with a unique assembly of carbonyl and olefinic moieties, are versatile intermediates in many chemical, medicinal, and biological processes.^[13] We envisioned that copper-catalyzed enantioselective 1,6-conjugate boration of *p*-QMs could deliver nonracemic *gem*-diarylmethine boronates. (Scheme 1b). However, *p*-QMs have been less-well studied in enantioselective catalysis, and to date, only three organocatalysis cases, including anionic polymerizations,^[14] phase-transfer catalysis,^[15] and 1,6-conjugate addition of enamines,^[16] have been reported. Herein, we report the first copper-catalyzed asymmetric 1,6-conjugate addition of B₂(pin)₂ to *p*-QMs, and stereoselective conversion of the chiral *gem*-diarylmethine boronates into enantioenriched triarylmethanes.

To test the feasibility of the 1,6-addition, we examined the reaction of the *p*-QM **1a** with B₂(pin)₂ using a P-diisopropyl sulfoxide phosphine^[17] (SOP; **L1**), and systematically screened reaction conditions, such as copper salts, solvents, additives, and bases (Table 1; see the Supporting Information for details). With 10 mol % of CuCl, **L1**, and NaOtBu, the reaction proceeded smoothly at –20 °C in the presence of 2 equivalents of MeOH and toluene as the solvent. **1a** was consumed within 15 hours and the 1,6-conjugate adduct, the dibenzylic boronic ester **2a**, was afforded in excellent yield with a 93.5:6.5 e.r. (entry 1). By employing MeOK as the base, the reaction proceeded within 30 min and gave **2a** with 95:5 e.r. (entry 2). Ligand screening revealed that P-substituents of SOPs have dramatic effects on the enantioselectivity. For instance, P-diethyl and P-dicyclohexyl SOPs (**L2** and **L3**) show much worse enantiocontrol than **L1**, and **L4** gave nearly racemic dibenzylic boronic ester (entries 3–5). The commercially available ligand (*R*)-BINAP was chosen to evaluate this transformation, and modest enantioselectivity was afforded (entry 6). Furthermore, the modified P-diisopropyl **L6** was

[*] Y. Z. Lou, Prof. Dr. P. Cao, T. Jia, Y. Zhang, Dr. M. Wang, Prof. Dr. J. Liao
Chengdu Institute of Biology, Chinese Academy of Sciences
Chengdu 610041 (China)
and
University of Chinese Academy of Sciences
Beijing 100049 (China)
E-mail: jliao@cib.ac.cn
Prof. Dr. J. Liao
College of Chemical Engineering, Sichuan University
Chengdu 610065 (China)

Supporting information for this article is available on the WWW under <http://dx.doi.org/10.1002/anie.201505926>.

Table 1: Screening of reaction conditions.^[a]


$\text{1a} + \text{B}_2(\text{pin})_2 \xrightarrow[\text{base, toluene, additive}]{\text{CuCl, L}^*} \text{2a}$

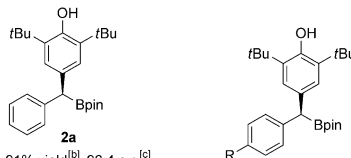
Ligands: L1 (R = *i*Pr), L2 (R = Et), L3 (R = Cy), L4 (R = Ph), L5 (R' = *i*Pr), L6 (R' = Cy), and (R)-BINAP.

Entry	Ligand	Base	<i>t</i> [h]	Yield [%] ^[b]	e.r. ^[c]
1	L1	NaOtBu	15	98	93.5:6.5
2	L1	MeOK	0.5	96	95:5
3	L2	MeOK	0.5	86	70:30
4	L3	MeOK	0.5	74	89:11
5	L4	MeOK	0.5	85	55:45
6	(R)-BINAP	MeOK	0.5	89	75.5:24.5
7	L5	MeOK	0.5	85	95:5
8	L6	MeOK	0.5	94	95.5:4.5
9 ^[d]	L6	MeOK	4.5	91	96:4
10 ^[e]	L6	MeOK	15	86	95:5
11 ^[f]	L6	MeOK	48	97	94:6
			(65)	(> 99.5:0.5)	

[a] Conducted with **1a** (0.2 mmol), $\text{B}_2(\text{pin})_2$ (0.3 mmol), CuCl (10 mol %), ligand (11 mol %), MeOK (12 mol %), MeOH (0.4 mmol) in toluene (1.0 mL) at -20°C . [b] Yield of isolated product. [c] Determined by HPLC. [d] Conducted with 5.0 mol % of CuCl at -40°C . [e] Conducted with 2.5 mol % of CuCl at -40°C . [f] The reaction was performed in the presence of 1.60 g (8.1 mmol) of **1a** and 0.1 mol % of CuCl at 0°C . Data within parentheses is that obtained after one crystallization.

used and it led to a slight enantiomeric excess increase (entry 8). Reaction temperature and catalyst loading were also optimized (entries 9 and 10). At -40°C and with 5 mol % of catalyst, **1a** reacted completely with $\text{B}_2(\text{pin})_2$ within 4.5 hours to furnish **2a** in 91% yield and 96:4 e.r. The reaction can also be run on a gram scale (1.60 g) with 0.1 mol % catalyst at 0°C , thus delivering **2a** in excellent yield (2.23 g, 97%) and good enantioselectivity (94:6 e.r.). After one recrystallization from *n*-hexane, enantiopure **2a** (>99.5:0.5 e.r.) can be obtained in 65% yield (1.50 g; entry 11).

Under the optimized reaction conditions, we examined the scope of the reaction. As shown in Table 2, a series of stable *p*-QMs with electron-neutral, electron-deficient, and electron-rich aryls reacted with $\text{B}_2(\text{pin})_2$ to provide chiral *gem*-diarylmethine boronates (**2a–m**) in good yields with generally high levels of enantiopurities. Multisubstituted (**2n** and **2o**), polycyclic (**2p**), and heterocyclic (**2s**) aryls were tolerated in this reaction and resulted in excellent yields with high enantioselectivities. *o*-Methoxy- and *o*-methyl-substituted *p*-QMs provided **2q** and **2r**, respectively, with decreased e.r. values (83:17 and 75:25), probably because of the steric effects. In addition, replacement of the bulky *tert*-butyl group by methyl (**2t**), isopropyl (**2u**), and phenyl (**2v**) groups did not

Table 2: Scope with respect to the *p*-QMs.^[a]


$\text{1} + \text{B}_2(\text{pin})_2 \xrightarrow[\text{MeOK, MeOH}]{\text{CuCl}} \text{2}$

Substituents: **2b** R = F, **2c** R = Cl, **2d** R = Br, **2e** R = Me, **2f** R = *i*Pr, **2g** R = *t*Bu, **2h** R = MeO, **2i** R = BnO, **2j** R = F, **2k** R = Br, **2l** R = Me, **2m** R = Me, **2n** R = Me, **2o** R = Me, **2p** R = Me, **2q** R = Me, **2r** R = Me, **2s** R = Me, **2t** R = Me, **2u** R = Me, **2v** R = Me.

Entry	Yield [%]	e.r.
2a	91% yield ^[b] , 96:4 e.r. ^[c]	
2b	90% yield, 97:3 e.r.	
2c	84% yield, 96.5:3.5 e.r.	
2d	87% yield, 95:5 e.r.	
2e	90% yield, 97:3 e.r.	
2f	98% yield, 95:5 e.r.	
2g	97% yield, 94.5:5.5 e.r.	
2h	88% yield, 96.5:3.5 e.r.	
2i	94% yield, 97:3 e.r.	
2j	94% yield, 95:5 e.r.	
2k	93% yield, 94.5:5.5 e.r.	
2l	92% yield, 95:5 e.r.	
2m	95% yield, 95:5 e.r.	
2n	91% yield, 97.5:2.5 e.r.	
2o	93% yield, 92.5:7.5 e.r.	
2p	97% yield, 95:5 e.r.	
2q	71% yield, 83:17 e.r.	
2r	78% yield, 75:25 e.r.	
2s	96% yield, 98.5:1.5 e.r.	
2t	89% yield, 75:25 e.r.	
2u	93% yield, 84:16 e.r.	
2v	83% yield, 93:7 e.r.	

[a] Conducted with **1** (0.2 mmol), $\text{B}_2(\text{pin})_2$ (0.3 mmol), CuCl (5 mol %), ligand (5.5 mol %), MeOK (6 mol %), MeOH (0.4 mmol) in toluene (1.0 mL) at -40°C . [b] Yield of isolated product. [c] Determined by chiral HPLC. Thermal ellipsoids shown at 50% probability.

lead to better enantioselectivities. The stereocenter of the generated dibenzylic boronic esters was assigned to be *S* by X-ray crystallographic analysis of **2a**.^[18] Accordingly, a model was proposed to rationalize this stereochemical outcome. As shown in Figure 1, the planar **1a** approaches the (SOP)Cu/Bpin species from *Si*-face (**A**) to avoid steric interactions between the phenyl ring of **1a** and *tert*-butyl sulfinyl group of **L6** (the steric hindrance is shown in **B**), thus (*S*)-**2a** was obtained in a large excess. For *o*-methoxy- or *o*-methyl-

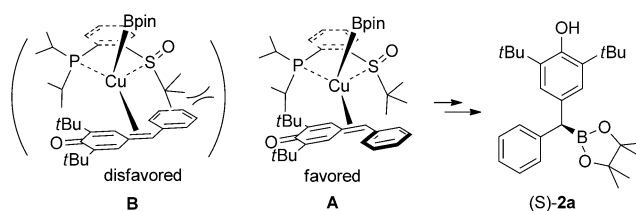

Figure 1. Stereochemical models of 1,6-boration.

Table 3: Palladium-catalyzed cross-coupling of *gem*-diarylmethine trifluoroborates with ArOTf.^[a]

3aa 74% yield^[b], 98:2 e.r.^[c] 97% es^[d]	3ab 78% yield, 98:2 e.r. 97% es	3ac 53% yield, 96:4 e.r. 93% es
3ad 66% yield, 93:7 e.r. 87% es	3ae 70% yield, 97.5:2.5 e.r. 96% es	3af 80% yield, 98:2 e.r. 97% es
3ag 57% yield, 96.5:3.5 e.r. 94% es	3ah 50% yield, 98:2 e.r. 97% es	3bb 62% yield, 92:8 e.r. 89% es
3be 71% yield, 93.5:6.5 e.r. 93% es	3eb 68% yield, 95.5:4.5 e.r. 97% es	3ee 65% yield, 95.5:4.5 e.r. 97% es

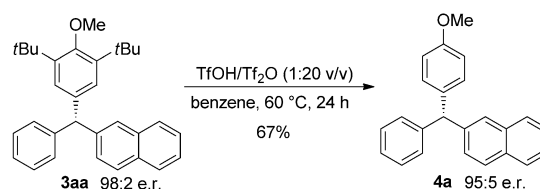
[a] Reaction condition of the cross-coupling reaction: **2** (0.07 mmol, e.r. > 99.5:0.5 for **2a**), ArOTf (0.07 mmol), Pd(OAc)₂ (10 mol %), PCy₃ (20 mol %), K₂CO₃ (0.2 mol) in toluene/H₂O (1.1 mL, v/v = 10:1) at 60 °C for 15 h. [b] Yield of isolated product. [c] Determined by chiral HPLC. [d] es = $ee_{\text{Product}}/ee_{\text{S.M.}} \times 100\%$. Tf = trifluoromethanesulfonyl, THF = tetrahydrofuran.

substituted *p*-QMs, an assumed orthogonal orientation of cyclohexadiene ring with the sterically bulky aryl ring might render the enantioface discrimination elusive, thus resulting in moderate e.r. values of 1,6-borylative adducts.

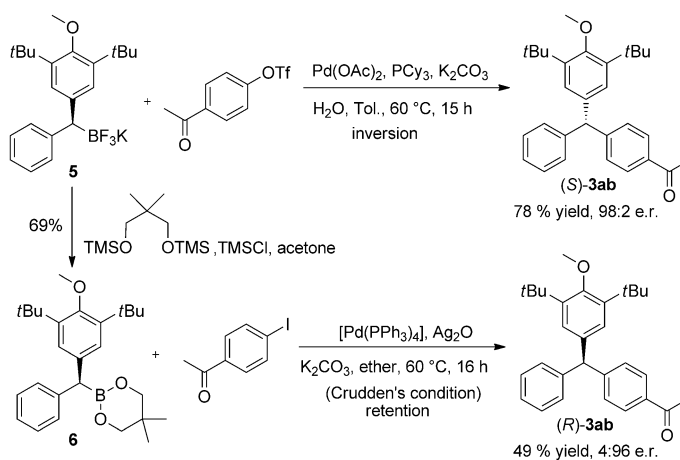
To demonstrate the utility of this enantioselective 1,6-boration in asymmetric synthesis, we sought to transform the 1,6-adducts into enantioenriched triaryls by stereospecific Suzuki–Miyaura cross-coupling reactions.^[19] Firstly, all attempts at the palladium-catalyzed cross-coupling of **2a** with aryl halides or pseudohalides failed, and is probably a result of the facile decomposition of the dibenzyl palladium intermediate when bearing a phenolic hydroxy group. Fortunately, we can

readily transfer boronic esters to potassium trifluoroborates and then protect the phenols with a methyl group in a one-pot procedure in satisfactory yields (Table 3). The new dibenzyl trifluoroborates can be readily coupled with aryltrifluoromethanesulfonates (ArOTf) in the presence of catalytic amounts of Pd(OAc)₂ and PCy₃ under milder reaction conditions. By using this method, a series of enantioenriched triarylmethanes were successfully constructed with high enantiospecificity.^[20] Electronic properties of ArOTf have a slight influence on yield as well as specificity, and the cross-coupling procedure was tolerated for a wide range of functional groups, such as carbonyl (**3ab**), formyl (**3ac**), nitro (**3ad**), and alkoxy (**3ae** and **3af**). This method also enables efficient synthesis of four different functionalized triarylmethanes (**3bb**, **3be**, **3eb**, **3ee**). The current method, using ArOTf as electrophiles, represents a practical cross-coupling technique owing to the availability of phenols and their lack of toxicity compared to aryl halides.^[7,19b,21,22]

By exposing **3aa** to a mixture of Tf₂O and TfOH, the enantioenriched triarylmethane was readily converted into the de-*tert*-butylated triarylmethane **4a** in 67% yield with a slight loss of stereopurity (Scheme 2). The absolute configuration of **4a** was determined to be *R* by comparison of the optical rotation with the literature value.^[20c] It is noteworthy that both enantiomers (*R* and *S*) of the triarylmethanes can be obtained from the 1,6-adducts **2**. Thus whilst coupling of the trifluoroborate **5** with ArOTf occurred with inversion,^[23] the opposite stereochemistry (retention) was observed when coupling the dibenzyl neopentyl glycol boronic ester **6** with aryl iodide.^[7] (Scheme 3).



Scheme 2. De-*tert*-butylation of triarylmethane.



Scheme 3. Stereospecific cross-coupling experiments. TMS = trimethylsilyl.

In summary, we have demonstrated that copper-catalyzed asymmetric 1,6-addition of $B_2(\text{pin})_2$ to the *p*-QMs proceeds with excellent yield and good to excellent enantioselectivities. For the first time, enantioenriched *gem*-diarylmethine boronates were prepared by a catalytic asymmetric method. Subsequently, the derived chiral potassium trifluoroborates were converted into enantioenriched triarylmethanes in good yield and with high enantiospecificity. Additional asymmetric transformations of *gem*-diarylmethine boronates are under way in our laboratory.

Acknowledgements

We thank the NSFC (No. 21472184, 21272226, 21202160, and 21402186) for financial support.

Keywords: asymmetric catalysis · boron · copper · cross-coupling · synthetic methods

How to cite: *Angew. Chem. Int. Ed.* **2015**, *54*, 12134–12138
Angew. Chem. **2015**, *127*, 12302–12306

- [1] a) S. J. Baker, J. W. Tomshob, S. J. Benkovic, *Chem. Soc. Rev.* **2011**, *40*, 4279; b) S. J. Baker, C. Z. Ding, T. Akama, Y.-K. Zhang, V. Hernandez, Y. Xia, *Future Med. Chem.* **2009**, *1*, 1275; c) W. Yang, X. Gao, B. Wang, *Med. Res. Rev.* **2003**, *23*, 346; d) M. P. Groziak, *Am. J. Ther.* **2001**, *8*, 321.
- [2] Reviews: a) B. W. Glasspoole, C. M. Crudden, *Nat. Chem.* **2011**, *3*, 912; b) E. Swift, E. R. Jarvo, *Tetrahedron* **2013**, *69*, 5799; c) D. Leonori, V. K. Aggarwal, *Angew. Chem. Int. Ed.* **2015**, *54*, 1082; *Angew. Chem.* **2015**, *127*, 1096.
- [3] Reviews: a) D. F. Duxbury, *Chem. Rev.* **1993**, *93*, 381; b) M. Beija, C. A. M. Afonso, J. M. G. Martinho, *Chem. Soc. Rev.* **2009**, *38*, 2410; c) V. Nair, S. Thomas, S. C. Mathew, K. G. Abhilash, *Tetrahedron* **2006**, *62*, 6731; d) M. S. Shchepinov, V. A. Korshun, *Chem. Soc. Rev.* **2003**, *32*, 170; Selective literatures: e) M. K. Parai, G. Panda, V. Chaturvedi, Y. K. Manju, S. Sinha, *Bioorg. Med. Chem. Lett.* **2008**, *18*, 289; f) R. A. Al-Qawasmeh, Y. Lee, M. Cao, X. Gu, A. Vassilikos, J. A. Wright, A. Young, *Bioorg. Med. Chem. Lett.* **2004**, *14*, 347; g) R. Palchadhuri, V. Nesterenko, P. J. Hergenrother, *J. Am. Chem. Soc.* **2008**, *130*, 10274; h) G. Panda, Shagufta, J. K. Mishra, V. Chaturvedi, A. K. Srivastava, R. Srivastava, B. S. Srivastava, *Bioorg. Med. Chem.* **2004**, *12*, 5269; i) Shagufta, A. K. Srivastava, R. Sharma, R. Mishra, A. K. Balapure, P. S. R. Murthy, G. Panda, *Bioorg. Med. Chem.* **2006**, *14*, 1497.
- [4] a) A. L. McRae, K. T. Brady, *Expert Opin. Pharmacother.* **2001**, *2*, 883; b) C. J. Hills, S. A. Winter, J. A. Balfour, *Drugs* **1998**, *55*, 813; c) K. P. Bogeso, A. V. Christensen, J. Hyttel, T. Liljefors, *J. Med. Chem.* **1985**, *28*, 1817; d) J. Hyttel, J. J. Larsen, *J. Neurochem.* **1985**, *44*, 1615; e) M. Gordaliza, P. A. García, J. M. Miguel del Corral, M. A. Castro, M. A. Gómez-Zurita, *Toxicol.* **2004**, *44*, 441.
- [5] a) J. L. Stymiest, V. Bagutski, R. M. French, V. K. Aggarwal, *Nature* **2008**, *456*, 778; b) A. Ros, V. K. Aggarwal, *Angew. Chem. Int. Ed.* **2009**, *48*, 6289; *Angew. Chem.* **2009**, *121*, 6407; c) S. Nave, R. P. Sonawane, T. G. Elford, V. K. Aggarwal, *J. Am. Chem. Soc.* **2010**, *132*, 17096; d) S. Roesner, J. M. Casatejada, T. G. Elford, R. P. Sonawane, V. K. Aggarwal, *Org. Lett.* **2011**, *13*, 5740; e) S. Roesner, V. K. Aggarwal, *Can. J. Chem.* **2012**, *90*, 965; f) A. Bonet, M. Odachowski, D. Leonori, S. Essafi, V. K. Aggarwal, *Nat. Chem.* **2014**, *6*, 584.
- [6] For seminal examples, see: a) E. Beckmann, V. Desai, D. Hoppe, *Synlett* **2004**, 2275; b) E. Beckmann, D. Hoppe, *Synthesis* **2005**, 217; c) M. J. McGrath, P. O'Brien, *Synthesis* **2006**, 2233; d) J. L. Stymiest, G. Dutheil, A. Mahmood, V. K. Aggarwal, *Angew. Chem. Int. Ed.* **2007**, *46*, 7491; *Angew. Chem.* **2007**, *119*, 7635.
- [7] S. C. Matthew, B. W. Glasspoole, P. Eisenberger, C. M. Crudden, *J. Am. Chem. Soc.* **2014**, *136*, 5828.
- [8] For selected copper-catalyzed conjugated boronates, see: a) K. Takahashi, T. Ishiyama, N. Miyaura, *Chem. Lett.* **2000**, 982; b) H. Ito, H. Yamanaka, J.-I. Tateiwa, A. Hosomi, *Tetrahedron Lett.* **2000**, *41*, 6821; c) S. Mun, J.-E. Lee, J. Yun, *Org. Lett.* **2006**, *8*, 4887; d) J.-E. Lee, J. Yun, *Angew. Chem. Int. Ed.* **2008**, *47*, 145; *Angew. Chem.* **2008**, *120*, 151; e) V. Lillo, A. Prieto, A. Bonet, M. M. Díaz-Requejo, J. Ramírez, P. J. Pérez, E. Fernández, *Organometallics* **2009**, *28*, 659; f) J. C. H. Lee, R. McDonald, D. G. Hall, *Nat. Chem.* **2011**, *3*, 894; g) I. Ibrahim, P. Breistein, A. Córdova, *Angew. Chem. Int. Ed.* **2011**, *50*, 12036; *Angew. Chem.* **2011**, *123*, 12242; h) A. R. Burns, J. S. Gonzalez, H. W. Lam, *Angew. Chem. Int. Ed.* **2012**, *51*, 10827; *Angew. Chem.* **2012**, *124*, 10985; i) Y. Luo, I. D. Roy, A. G. E. Madec, H. W. Lam, *Angew. Chem. Int. Ed.* **2014**, *53*, 4186; *Angew. Chem.* **2014**, *126*, 4270; j) V. Hornillos, C. Vila, E. Otten, B. L. Feringa, *Angew. Chem. Int. Ed.* **2015**, *54*, 7867; *Angew. Chem.* **2015**, *127*, 7978; For rhodium-, nickel-, and palladium-catalyzed asymmetric conjugated boronates, see: k) T. Shiomi, T. Adachi, K. Toribatake, L. Zhou, H. Nishiyama, *Chem. Commun.* **2009**, 5987; l) V. Lillo, M. J. Geier, S. A. Westcott, E. Fernández, *Org. Biomol. Chem.* **2009**, *7*, 4674; m) K. Toribatake, L. Zhou, A. Tsuruta, H. Nishiyama, *Tetrahedron* **2013**, *69*, 3551.
- [9] a) A. Bonet, H. Gulyás, E. Fernández, *Angew. Chem. Int. Ed.* **2010**, *49*, 5130; *Angew. Chem.* **2010**, *122*, 5256; b) I. Ibrahim, P. Breistein, A. Cordova, *Chem. Eur. J.* **2012**, *18*, 5175; c) H. Wu, S. Radomkit, J. M. O'Brien, A. H. Hoveyda, *J. Am. Chem. Soc.* **2012**, *134*, 8277; d) S. Radomkit, A. H. Hoveyda, *Angew. Chem. Int. Ed.* **2014**, *53*, 3387; *Angew. Chem.* **2014**, *126*, 3455.
- [10] Selected reviews and highlights: a) E. Hartmann, D. J. Vyas, M. Oestreich, *Chem. Commun.* **2011**, 47, 7917; b) A. D. J. Calow, A. Whiting, *Org. Biomol. Chem.* **2012**, *10*, 5485; c) J. A. Schiffrer, K. Mütther, M. Oestreich, *Angew. Chem. Int. Ed.* **2010**, *49*, 1194; *Angew. Chem.* **2010**, *122*, 1214; d) L. Mantilli, C. Mazet, *ChemCatChem* **2010**, *2*, 501.
- [11] For selected recent examples, see: a) I.-H. Chen, L. Yin, W. Itano, M. Kanai, M. Shibasaki, *J. Am. Chem. Soc.* **2009**, *131*, 11664; b) J. M. O'Brien, K.-s. Lee, A. H. Hoveyda, *J. Am. Chem. Soc.* **2010**, *132*, 10630; c) D. Hirsch-Weil, K. A. Abboud, S. Hong, *Chem. Commun.* **2010**, 46, 7525; d) A. L. Moure, R. Gomez Arayas, J. C. Carretero, *Chem. Commun.* **2011**, 47, 6701; e) E. Hartmann, M. Oestreich, *Org. Lett.* **2012**, *14*, 2406; f) L. Zhao, Y. Ma, W. Duan, F. He, J. Chen, C. Song, *Org. Lett.* **2012**, *14*, 5780; g) A. D. J. Calow, A. S. Batsanov, E. Fernández, C. Sol, A. Whiting, *Chem. Commun.* **2012**, 48, 11401; h) S. Kobayashi, P. Xu, T. Endo, M. Ueno, T. Kitanosono, *Angew. Chem. Int. Ed.* **2012**, *51*, 12763; *Angew. Chem.* **2012**, *124*, 12935; i) T. Iwai, Y. Akiyama, M. Sawamura, *Tetrahedron: Asymmetry* **2013**, *24*, 729; j) Z.-T. He, Y.-S. Zhao, P. Tian, C.-C. Wang, H.-Q. Dong, G.-Q. Lin, *Org. Lett.* **2014**, *16*, 1426; k) C. T. Check, K. P. Jang, C. B. Schwamb, A. S. Wong, M. H. Wang, K. A. Scheidt, *Angew. Chem. Int. Ed.* **2015**, *54*, 4264; *Angew. Chem.* **2015**, *127*, 4338; l) J.-B. Xie, S. Lin, J. Luo, J. Wu, T. R. Winna, G. Li, *Org. Chem. Front.* **2015**, *2*, 42.
- [12] For reviews on the chemistry of *p*-QMs, see: a) A. B. Turner, *Q. Rev. Chem. Soc.* **1964**, *18*, 347; b) "Quinone Methides": H.-U. Wagner, R. Gompper in *The Chemistry of the Quinonoid Compounds*, Vol. 2 (Ed.: S. Patai), Wiley, New York, **1974**, chap. 18, pp. 1145; c) M. G. Peter, *Angew. Chem. Int. Ed. Engl.* **1989**, *28*, 555; *Angew. Chem.* **1989**, *101*, 572; d) T. Itoh, *Prog. Polym. Sci.* **2001**, *26*, 1019; e) *Quinone Methides* (Ed.: S. E.

- Rokita), Wiley, Hoboken, **2009**; f) M. M. Toteva, J. P. Richard, *Adv. Phys. Org. Chem.* **2011**, 45, 39.
- [13] a) A. A. Larsen, *Nature* **1969**, 224, 25; b) D. Hamels, P. M. Dansette, E. A. Hillard, S. Top, A. Vessi res, P. Herson, G. Jaouen, D. Mansuy, *Angew. Chem. Int. Ed.* **2009**, 48, 9124; *Angew. Chem.* **2009**, 121, 9288; c) G. B. Messiano, T. da Silva, I. R. Nascimento, L. M. X. Lopes, *Phytochemistry* **2009**, 70, 590; d) R. Dehn, Y. Katsuyama, A. Weber, K. Gerth, R. Jansen, H. Steinmetz, G. H fle, R. M ller, A. Kirschning, *Angew. Chem. Int. Ed.* **2011**, 50, 3882; *Angew. Chem.* **2011**, 123, 3968; e) C. Sridar, J. D'Agostino, P. F. Hollenberg, *Drug. Circ. Drug. Metab. Dispos.* **2012**, 40, 2280.
- [14] a) N. Nakagaki, T. Uno, M. Kubo, T. Itoh, *J. Polym. Sci. Part A* **2009**, 47, 5923; b) S. Lizuka, N. Nakagaki, T. Uno, M. Kubo, T. Itoh, *Macromolecules* **2010**, 43, 6962; c) R. Nakagawa, T. Uno, M. Kubo, T. Itoh, *Polym. Bull.* **2012**, 68, 1831; d) T. Nagai, T. Uno, M. Kubo, T. Itoh, *J. Polym. Sci. Part A* **2012**, 50, 466.
- [15] W. Chu, L. Zhang, X. Bao, X. Zhao, C. Zeng, J. Du, G. Zhang, F. Wang, X. Ma, C. Fan, *Angew. Chem. Int. Ed.* **2013**, 52, 9229; *Angew. Chem.* **2013**, 125, 9399.
- [16] L. Caruana, F. Kniep, T. K. Johansen, P. H. Poulsen, K. A. J rgensen, *J. Am. Chem. Soc.* **2014**, 136, 15929.
- [17] a) J. Chen, D. Li, H. Ma, L. Cun, J. Zhu, J. Deng, J. Liao, *Tetrahedron Lett.* **2008**, 49, 6921; b) T. Jia, P. Cao, D. Wang, Y. Lou, J. Liao, *Chem. Eur. J.* **2015**, 21, 4918; c) D. Wang, P. Cao, B. Wang, T. Jia, Y. Lou, M. Wang, J. Liao, *Org. Lett.* **2015**, 17, 2420.
- [18] CCDC 1415208 (**2a**) contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre.
- [19] a) V. Bagutski, A. Ros, V. K. Aggarwal, *Tetrahedron* **2009**, 65, 9956; b) G. A. Molander, S. R. Wisniewski, *J. Am. Chem. Soc.* **2012**, 134, 16856; c) G. A. Molander, W. Febo-Ayala, M. Ortega-Guerra, *J. Org. Chem.* **2008**, 73, 6000; d) G. A. Molander, J. Ham, B. Canturk, *Org. Lett.* **2007**, 9, 821; e) G. A. Molander, N. M. Ellis, *J. Org. Chem.* **2006**, 71, 7491.
- [20] For asymmetric synthesis of triarylmethanes, see: a) B.-F. Shi, N. Mangel, Y.-H. Zhang, J.-Q. Yu, *Angew. Chem. Int. Ed.* **2008**, 47, 4882; *Angew. Chem.* **2008**, 120, 4960; b) F.-L. Sun, X.-J. Zheng, Q. Gu, Q.-L. He, S.-L. You, *Eur. J. Org. Chem.* **2010**, 47; c) B. L. H. Taylor, M. R. Harris, E. R. Jarvo, *Angew. Chem. Int. Ed.* **2012**, 51, 7790; *Angew. Chem.* **2012**, 124, 7910; d) M. R. Harris, L. E. Hanna, M. A. Greene, C. E. Moore, E. R. Jarvo, *J. Am. Chem. Soc.* **2013**, 135, 3303; e) Q. Zhou, H. D. Srinivas, S. Dasgupta, M. P. Watson, *J. Am. Chem. Soc.* **2013**, 135, 3307; f) Y. Huang, T. Hayashi, *J. Am. Chem. Soc.* **2015**, 137, 7556.
- [21] a) D. Imao, B. W. Glasspoole, V. S. Laberge, C. M. Crudden, *J. Am. Chem. Soc.* **2009**, 131, 5024; b) L. Chausset-Boissarie, K. Ghazati, E. LaBine, J. L.-Y. Chen, V. K. Aggarwal, C. M. Crudden, *Chem. Eur. J.* **2013**, 19, 17698; c) B. M. Partridge, L. Chausset-Boissarie, M. Burns, A. P. Pulis, V. K. Aggarwal, *Angew. Chem. Int. Ed.* **2012**, 51, 11795; *Angew. Chem.* **2012**, 124, 11965; d) T. Ohmura, T. Awano, M. Sugimoto, *J. Am. Chem. Soc.* **2010**, 132, 13191.
- [22] a) P. B. D. d. L. Mare, *Electrophilic Halogenation*, Cambridge University, New York, **1976**; b) P. J. Stang, M. Hanack, L. R. Subramanian, *Synthesis* **1982**, 85.
- [23] Inversion stereochemistry of palladium-catalyzed cross-coupling was observed. See: L. Li, S. Zhao, A. Joshi-Pangu, M. Diane, M. R. Biscoe, *J. Am. Chem. Soc.* **2014**, 136, 14027. For a discussion on retention versus inversion in Suzuki–Miyaura cross-coupling see Ref. [2c].

Received: June 28, 2015

Revised: July 27, 2015

Published online: August 28, 2015